

Appendix 1 – Chemical Storage, Segregation and Labelling





	U225 Highly flammable	flachpoint below the maximum ambient temperature
	liquid and vapour.	of work area may be kept in a laboratory/ workshop.
	H226 Flammable liquid	
	and vapour	No more than 250 litres for other flammable liquids
		with a higher flashpoint of up to 55°C may be stored
		with a higher hashpoint of up to 55 C may be stored
		in a laboratory/ workshop.
		Flammable liquids should be returned to the fire
		resisting cabinet immediately after use 500ml
		working volume may be kent an open banch then
		working volume may be kept on open bench, then
		returned to the storage area overnight. Empty
		flammable containers should be stored in the same
		way as full containers until removed to the waste
		store
		Store.
		Flammable liquids with <u>secondary hazards</u> (e.g.
		toxic/ corrosive) should be stored in a separate fire
		resisting cabinet. If it can be safely stored within the
		flommable applied it should be an a constant shalf
		nammable cabinet, it should be on a separate shell,
		in a secondary container or drip tray; refer to the
		SDS.

HSA-10101-08a







					to undergo a strong exothermic decomposition even without the participation of oxygen (air). Examples include various azo compounds.
olids substance	Class 4.2 Pyrophoric liquids and solids	H250 Catches fire spontaneously if exposed to air.			In small quantities, these should be stored in a dedicated fire resisting cabinet, used for Class 4.2 chemical only (see SDS Section 14). Do not store them with flammable liquids. Class 4.2 includes pyrophoric liquids and solids and self-heating substances and mixtures
Class 4 – Flammable soli 4.2 Sponteniously combustible su	Class 4.2 Self-heating substances and mixtures	H251 Self-heating: may catch fire. H252 Self-heating in large quantities; may catch fire.	Fire resisting cabinet	Spontaneously combustible	A pyrophoric solids and liquids A pyrophoric liquid or solid is a substance which, even in small quantities, is liable to ignite within 5 minutes of coming into contact with air. Pyrophoric substances have packaging that is designed to exclude air. If air enters a damaged package the substance may start to burn at room temperature or when gently heated. Examples include yellow phosphorus and some metal alkyls. Self-heating substances and mixtures Oxidative self-heating substances may react with the air, and so raise the temperature to the point at which spontaneous combustion takes place. This is normally a slow process which can be controlled by restricting the pack size, limiting storage duration,

HSA-10101-08a

Version 3.0

Page 4 of 11





HSA-10101-08a

Version 3.0

Page 5 of 11



					reducing agents or near combustible materials (e.g. paper/ cardboard).
Class 5 – Oxidising substances 5.2 – Organic perosides	Class 5.2 Organic peroxides	H242 Heating may cause a fire.	Fire resisting cabinet	Organic peroxide	 Organic peroxides These should be stored in a dedicated fire resisting cabinet, used for Class 5.2 chemicals (see SDS Section 14). Minimise the quantity stored and contact your H&S Lead. Some organic peroxides require temperature control. The manufacturer should provide information about the shelf-life and instructions on verifying desensitization, where applicable. Organic peroxides* are a particularly reactive type of oxidising substance. They may be solids, liquids or pastes, and have one or more of the following properties: liable to explosive decomposition. burn rapidly and intensely even in the absence of oxygen. sensitive to impact or friction. react dangerously with other substances decompose at comparatively low temperatures and/or cause spontaneous ignition if spilt onto combustible material.
					flammable, corrosive and toxic materials.

Version 3.0









HSA-10101-08a

Version 3.0



H304 May be fatal if swallowed and enters	
airways H305 May be harmful if swallowed and enters airways	



Version 3.0

Page **9** of **11**



ses	Â		Corrosive cabinet	Corrosive (inorganic acid)	Corrosive substances These should be stored in a dedicated corrosive cabinet, used for Class 8 chemicals (see SDS Section 14). Inorganic acids, organic acids and alkalis/bases must be in separate cabinets. Corrosive liquids must not be stored above shoulder height.
Class 8 – Corrosive substanc	Class 8 Corrosive material	Corrosive Corrosive H290 May be corrosive to metals. H314 Causes severe skin burns and eye damage. H318 Causes serious eye damage.	Corrosive cabinet	Corrosive (organic acid)	 Hazardous substances may be classified as corrosive because they burn the skin on contact or burn the mucous membranes of the respiratory tract by inhalation. Corrosive substances can cause serious eye damage. Corrosive substances will react with incompatible materials e.g., unsuitable packaging or metals (including shelving that is not corrosion resistant). Leaking corrosive substances may damage the packaging of other dangerous substances, thus creating further leaks. Corrosive (inorganic acid) / Corrosive (organic acid), have a pH less than 7. Separate storage cabinets are required to separate inorganic and organic acids; there is a risk of violent reactions if some inorganic and organic and organic

/ Corrosive ess than 7.

ts are required to anic acids; there is a some inorganic and ether.

HSA-10101-08a

Version 3.0

Page 10 of 11



			Nitric acid is corrosive, toxic and an oxidiser, it may be stored in the Corrosive (inorganic acid) cabinet inside a secondary container. Hydrofluoric acid must always be stored in a <u>dedicated</u> , locked cabinet, with access restricted to users who are trained in the safe handling of HF. Label: Hydrofluoric acid Corrosive Toxic
	Corrosive cabinet	Corrosive (alkali)	Corrosive (alkali / base), pH of greater than 7. Even although these materials are marked with a corrosive label, they must be stored separately from acids since any accidental mixing of the concentrated materials will generate large quantities of heat and fumes.

*A significant number of laboratory solvents can undergo autoxidation under normal storage conditions to form unstable and potentially dangerous peroxide by-products. They may be labelled EUH019 — 'May form explosive peroxides'. For essential information about their safe use and storage of refer to <u>Peroxide Forming Solvents</u>.

